

# Infra-red spectra of some selected minerals, rocks and products

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Forty-nine infra-red spectra ( $1600$  to  $400\text{ cm}^{-1}$ ) of rocks, minerals and products of industrial importance have been determined. These include mica, asbestos, apatite, calcite, feldspar, kyanite, slags, fly ash, etc. They have been classified into several groups based on structural parameters and chemical groups with pertinent discussions relating to the spectral characteristics of these materials for rapid identification and characterization.

## 1. Introduction

In this paper, forty-nine infra-red spectra of rocks, minerals and products which are of industrial importance have been presented. The spectra of the materials were obtained during the course of several projects carried out in our laboratory. These materials were of Indian origin and collected from all over the country at different times. The spectra appear to serve the purpose of identification of rocks, and minerals within certain limitations. In general, it has been found useful to analyse a large number of unknown samples with the help of this set of spectra within a short time before detailed analysis was undertaken by other methods such as X-ray diffraction, chemical analysis, etc. An attempt has been made here to correlate the spectra of the materials with respect to similarity in structural parameters, chemical groups, etc. and as such the spectra have been classified into thirteen groups for discussion. The spectral region of interest has been limited to the range  $1600$  to  $400\text{ cm}^{-1}$  and in some cases it has been extended up to  $3700\text{ cm}^{-1}$  for studying the hydroxyl stretching bands.

## 2. Experimental

The spectra of these materials were taken in potassium bromide pellets using a Parkin-Elmer Model 621 spectrometer. The spectra of mica samples were taken as film pressed on KBr pellets. The instrument was calibrated using polystyrene film as standard. The pressure applied for making the KBr pellets was around  $700\text{ kg cm}^{-2}$ . The

materials were finely powdered ( $-325$  mesh) in large quantities and representative spectra of samples were obtained in duplicate.

## 3. Absorption spectra

The fundamental modes of vibration which lead to absorption bands in the region under study are stretching, bending and lattice modes. The vibrational frequencies of these modes for the species under study vary considerably depending on parameters such as crystal forms, impurities in solid solution, etc. The presence of several phases of nearly similar absorption characteristics makes the study more complex, especially of rocks with variable composition.

### 3.1. Spectra of magnesite, calcite and aragonite

The carbonate rocks have three to four intense bands in this region (Fig. 1). The  $1420$  and  $876\text{ cm}^{-1}$  region bands are more or less unaltered in the spectra of these materials. The  $700\text{ cm}^{-1}$  band is characteristic for identification even in a mixture of these rocks because magnesite, calcite and aragonite absorb at  $748$ ,  $711$  and  $700\text{ cm}^{-1}$ , respectively. The aragonite sample is not pure but mixed with calcite. Further, aragonite has an extra characteristic band at  $1083\text{ cm}^{-1}$ .

#### 3.1.1. Spectra of dolomite, dolomitic limestone and kankar

The spectrum of dolomite is similar to that of calcite except for the  $700\text{ cm}^{-1}$  region band which

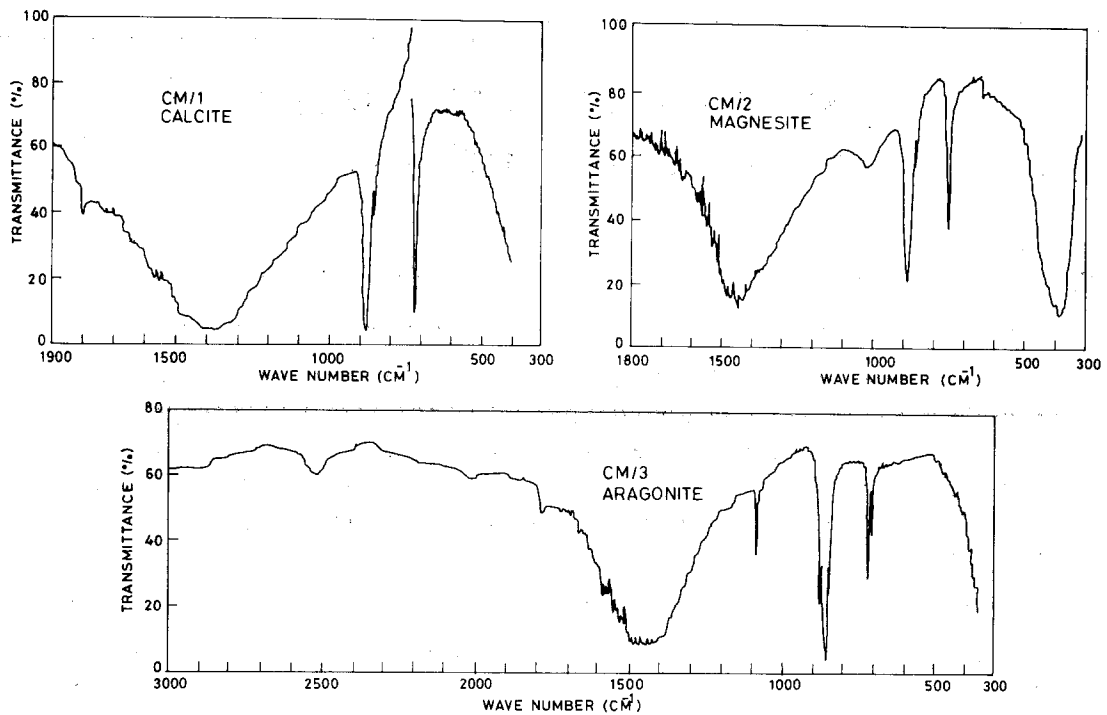


Figure 1 Spectra of magnesite, calcite and aragonite.

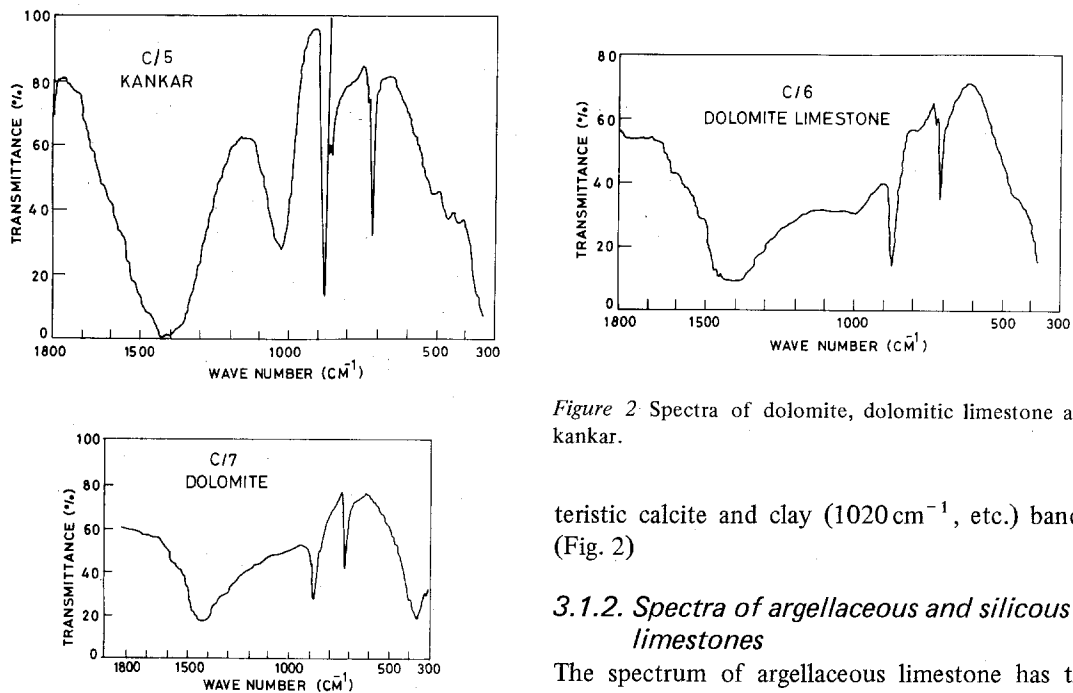


Figure 2 Spectra of dolomite, dolomitic limestone and kankar.

teristic calcite and clay ( $1020\text{ cm}^{-1}$ , etc.) bands. (Fig. 2)

### 3.1.2. Spectra of argillaceous and silicious limestones

The spectrum of argillaceous limestone has the usual calcite bands besides the bands at  $1000$ ,  $799$ ,  $510$  and  $462\text{ cm}^{-1}$ . The presence of clay is indicated. The spectrum of coral limestone indicates very little clay or silica. The spectra of silicious limestone and limestone with jasper show a high percentage of quartz as is evident from the

appears at  $727\text{ cm}^{-1}$  in dolomite. This bands has been used for estimation of dolomite in limestone [1]. The spectrum of kankar (limestone in the form of nodules used as a raw material for the manufacture of cement in India) shows charac-

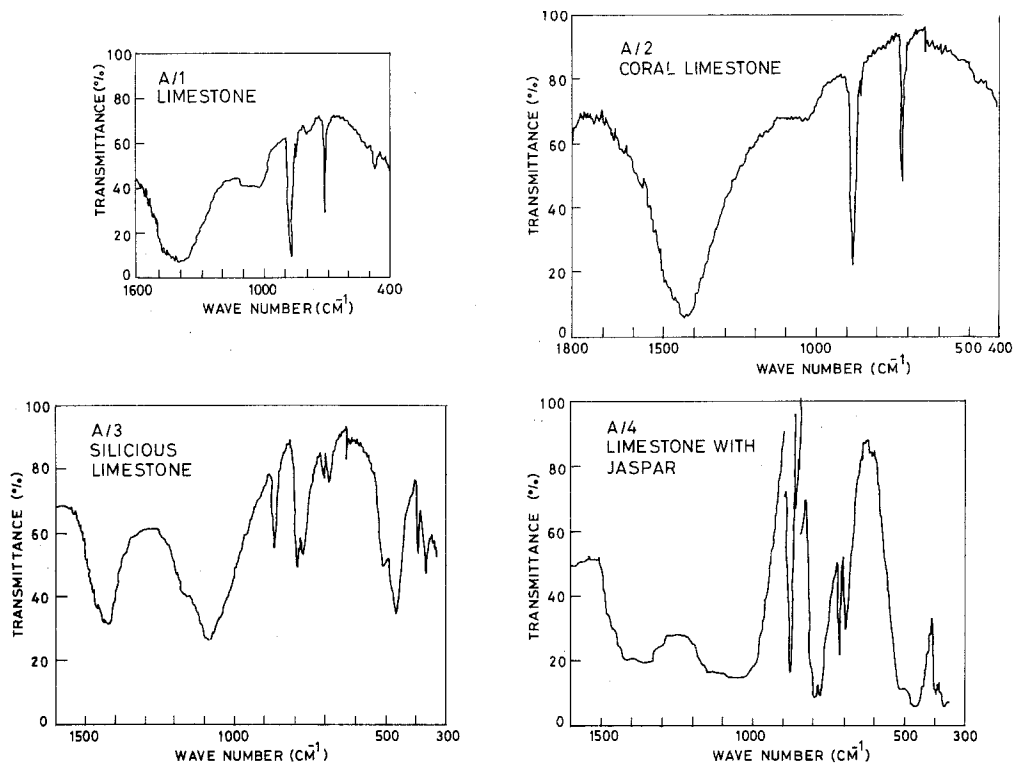


Figure 3 Spectra of argellaceous and silicious limestones.

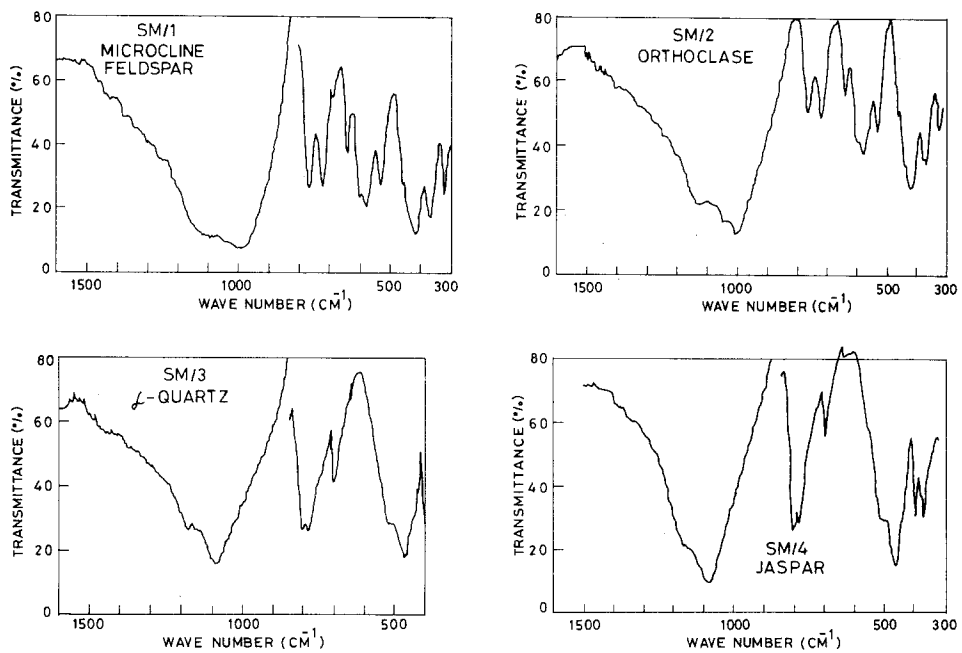


Figure 4 Spectra of feldspar, orthoclase, quartz and jasper.

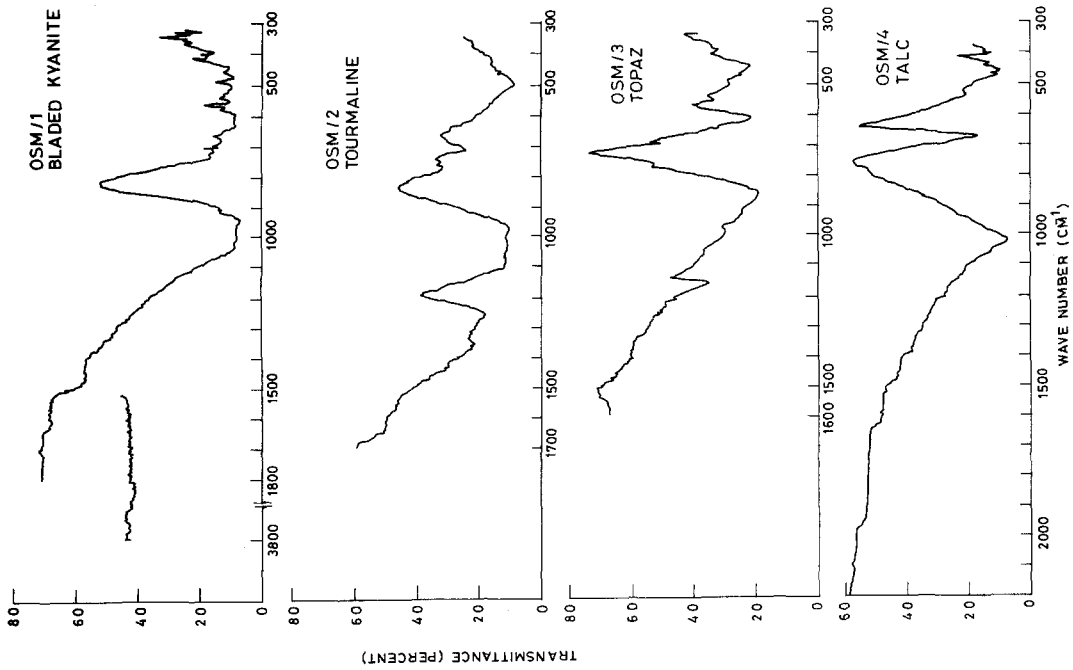


Figure 5 Spectra of tourmaline, kyanite, topaz and talc.

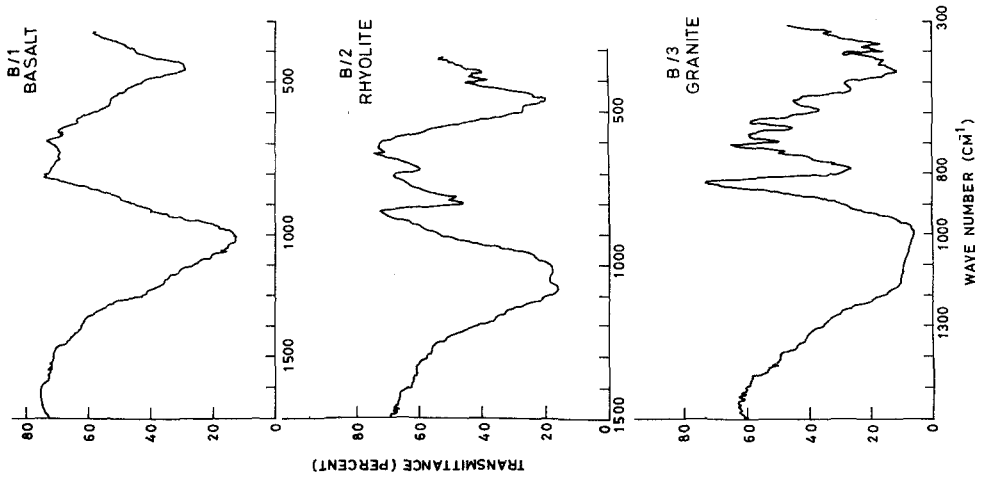


Figure 6 Spectra of rhyolite, granite and basalt.

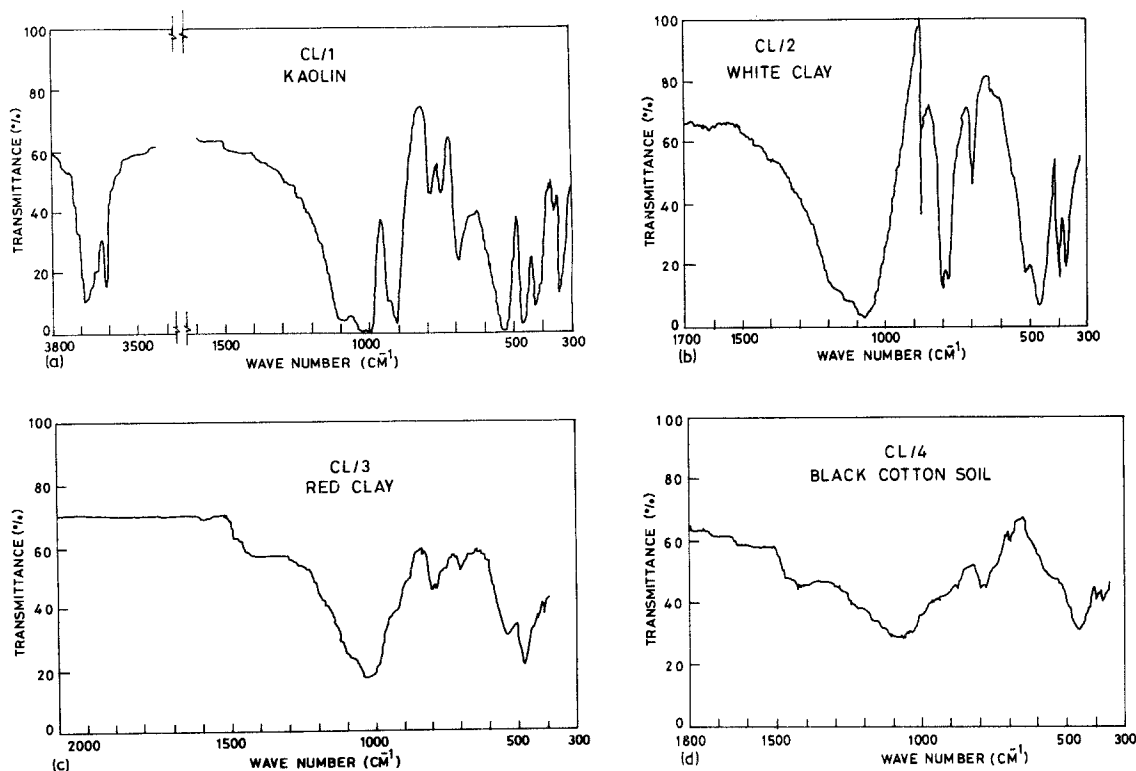


Figure 7 Spectra of kaolin and clays.

strong bands at 1165, 1090, 798, 775, 692, 515 and 465  $\text{cm}^{-1}$ . The quartz bands in the spectrum of jasper are broader especially at 1090  $\text{cm}^{-1}$  indicating the crypto-crystalline nature of jasper. (Fig. 3)

### 3.2. Spectra of feldspar, orthoclase, quartz and jasper

The spectrum of orthoclase is characterized by several bands at 1138, 1050, 1010, 772, 725, 645, 603, 584, 538, 468 and 425  $\text{cm}^{-1}$ . The spectrum of feldspar is quite similar to that of orthoclase except that the 1000  $\text{cm}^{-1}$  band is unresolved and the main peak head in this region shifts down to 980  $\text{cm}^{-1}$  with slight shifting of other bands. The spectra of  $\alpha$ -quartz and jasper are almost identical except some differences in band positions; for example, jasper has a band at 1165  $\text{cm}^{-1}$  while  $\alpha$ -quartz has this band at 1175  $\text{cm}^{-1}$ . (Fig. 4)

### 3.3. Spectra of tourmaline, kyanite, topaz and talc

The spectra of topaz and talc consist of few absorption bands. The spectrum of topaz has a characteristic medium intensity band at 1162  $\text{cm}^{-1}$ ,

a shoulder at 1000  $\text{cm}^{-1}$  and the main band head in this region at 865  $\text{cm}^{-1}$ , while the spectrum of talc has a weak shoulder at 1050  $\text{cm}^{-1}$  and the main band head is at 1015  $\text{cm}^{-1}$ . The other characteristic bands are at 670 and 615  $\text{cm}^{-1}$ . The hydroxyl stretching band appears at 3636  $\text{cm}^{-1}$  in the spectrum of topaz. Kyanite, being a nesosilicate (same as topaz) has a complex spectrum below 750  $\text{cm}^{-1}$  unlike that of topaz. The spectrum of tourmaline (cyclosilicate) has a characteristic doublet at 1300 and 1000  $\text{cm}^{-1}$ . The spectrum has some similarity with that of kaolinite below 800  $\text{cm}^{-1}$ . (Fig. 5)

### 3.4. Spectra of rhyolite, granite and basalt

The spectrum of granite indicates the presence of quartz (sharp band at 697  $\text{cm}^{-1}$ ) and orthoclase (at 650  $\text{cm}^{-1}$ ). The other regions are overlapped. The spectrum of basalt is simple since the principal constituents, albite and anorthite have  $\text{Si}_3\text{O}_8$  units and overlap in the same region. The spectrum of rhyolite shows the presence of quartz (bands at 800 and 780  $\text{cm}^{-1}$ ), but the presence of feldspar cannot be detected in its spectrum because of the amorphous nature of feldspar. (Fig. 6)

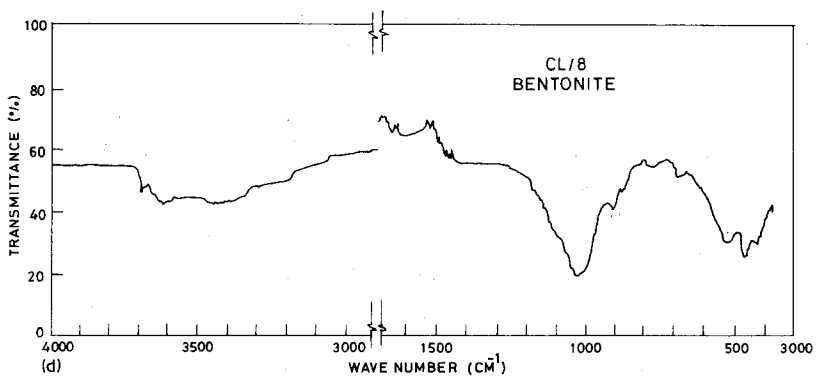
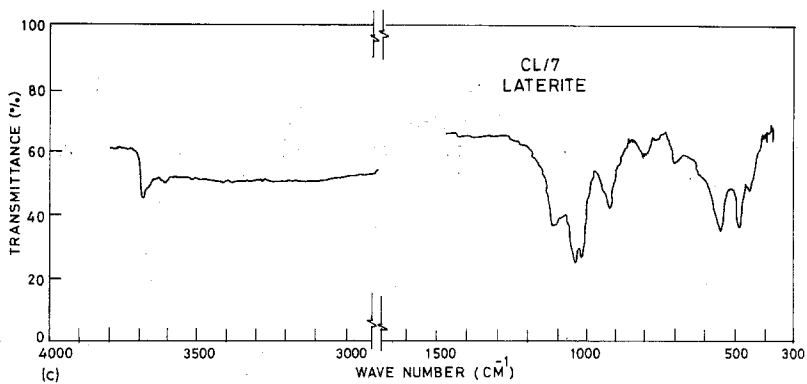
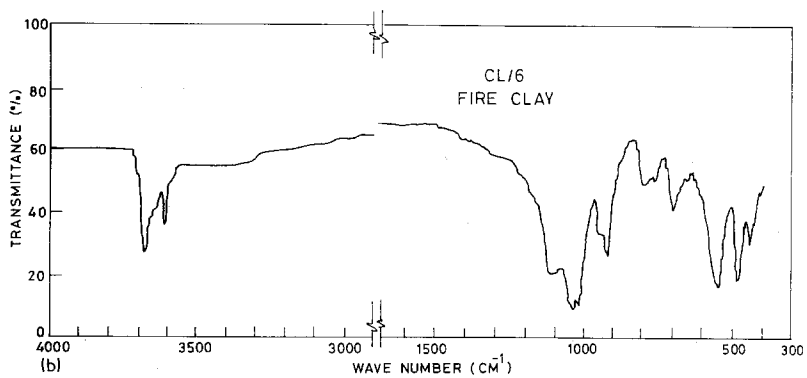
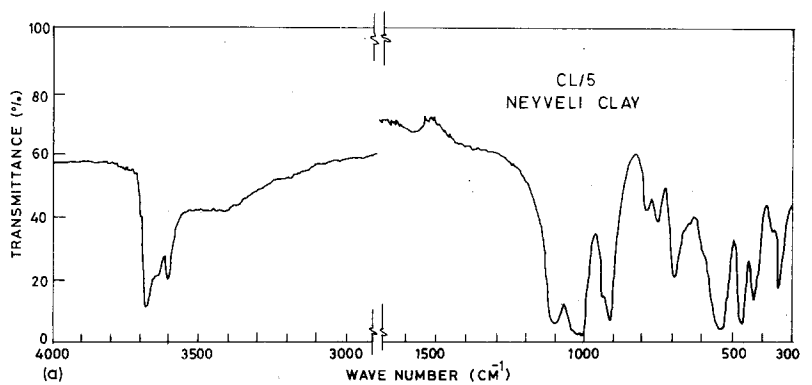
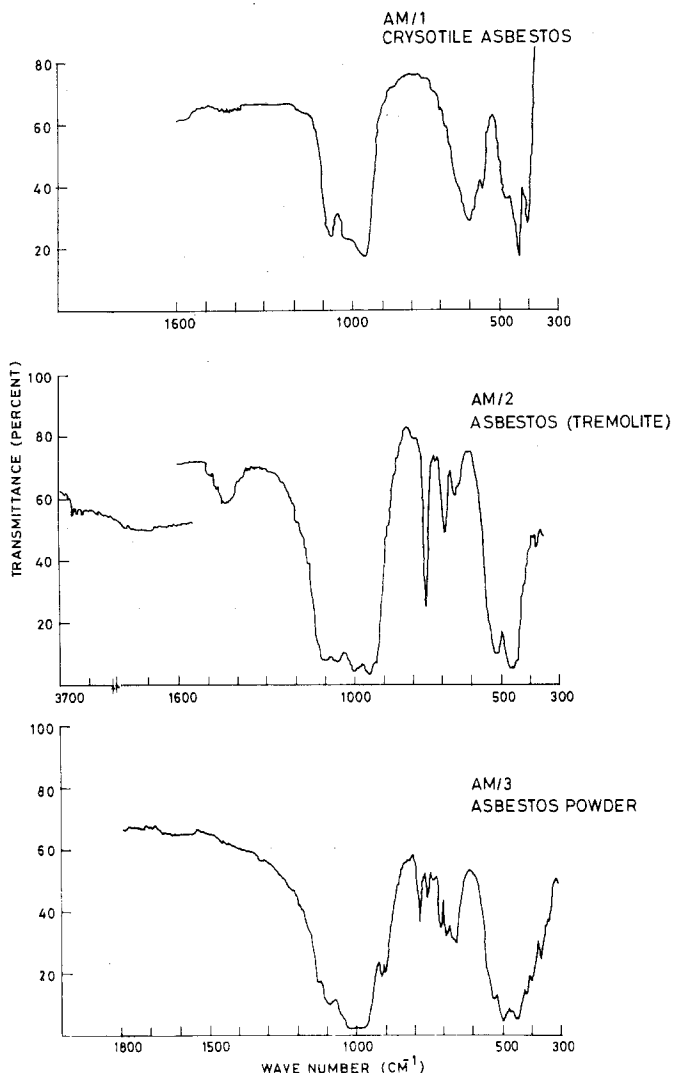


Figure 8 Spectra of bentonite, laterite, and other clays.

Figure 9 Spectra of chrysotile, tremolite and asbestos.



### 3.5. Spectra of kaolin, red clay, black cotton soil and white clay

The spectrum of kaolin is characterized by bands in the  $3600\text{ cm}^{-1}$  region (hydroxyl stretching); at  $1150$  to  $960\text{ cm}^{-1}$  (SiO stretching region) and below  $960\text{ cm}^{-1}$  (bending and lattice modes). The spectrum of white clay shows a broad and unresolved band in the Si-O stretching region and the presence of quartz ( $800$  and  $695\text{ cm}^{-1}$ ) is indicated. The Si-O stretching band appears at  $1025\text{ cm}^{-1}$  (red clay) and at  $1070\text{ cm}^{-1}$  (white clay). The spectrum of black cotton soil is diffuse and the soil sample is very poorly crystalline. Neyveli clay and fire clay are principally kaolinitic. The spectrum of bentonite indicates the presence of montmorillonite mineral, for example, bands at  $1102$ ,  $1030$ ,  $915$ ,  $521$  and  $467\text{ cm}^{-1}$  [2]. The

spectrum of laterite closely resembles those of kaolinite, goethite, gibbsite, etc. which are the constituents of laterite and have bands in the same region [3]. (Fig. 7 and 8)

### 3.6. Spectra of chrysotile, tremolite and asbestos

The spectra of chrysotile and tremolite are different. The former consists of  $\text{Si}_2\text{O}_5$  double-layers while the latter has  $\text{Si}_4\text{O}_{11}$  units in the structure. The  $1000\text{ cm}^{-1}$  Si-O stretching region has multiple bands in the spectrum of tremolite. The presence of carbonate in tremolite is shown by the weak band at  $1400\text{ cm}^{-1}$ . The spectrum of asbestos (poor) shows some bands ( $800$  to  $600\text{ cm}^{-1}$ ) common to tremolite, but the bands near  $900\text{ cm}^{-1}$  are not present in chrysotile or tremolite. (Fig. 9)

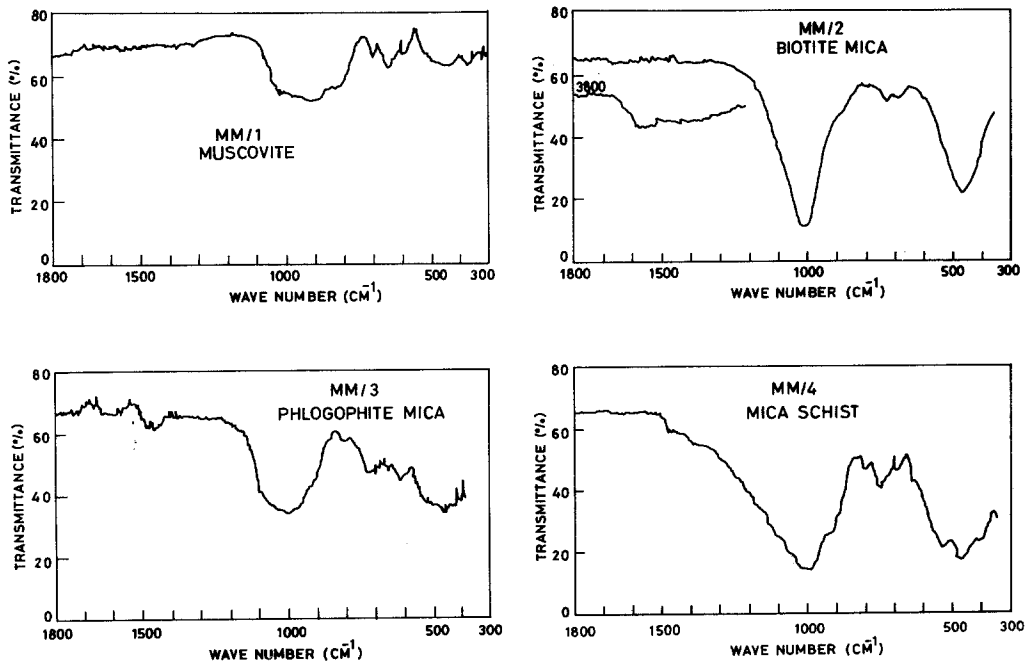


Figure 10 Spectra of biotite, muscovite mica, schist, and phlogophite.

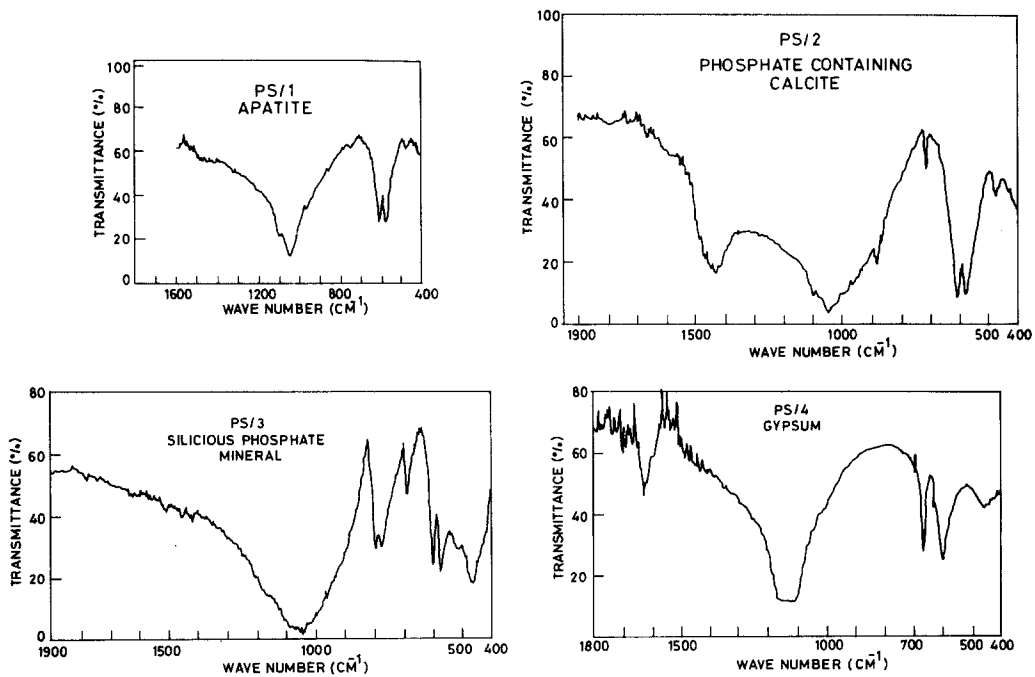


Figure 11 Spectra of apatites and gypsum.



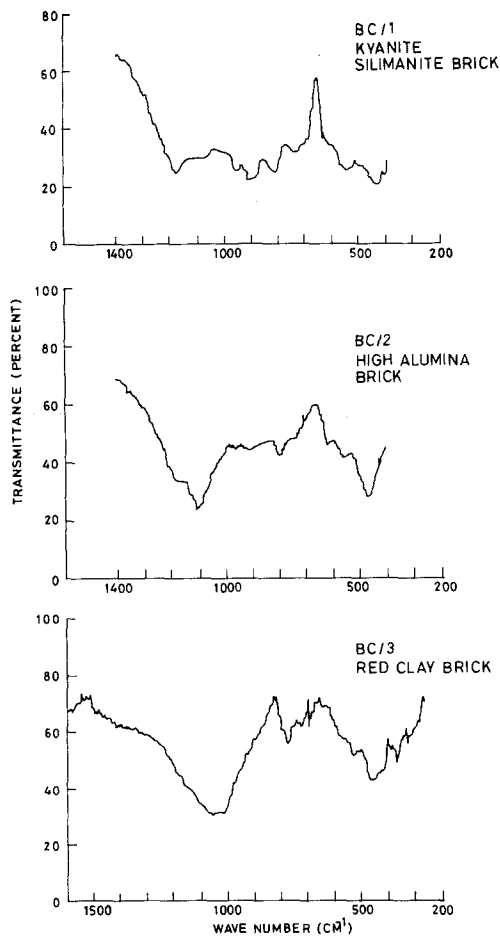


Figure 12 Spectra of bricks.

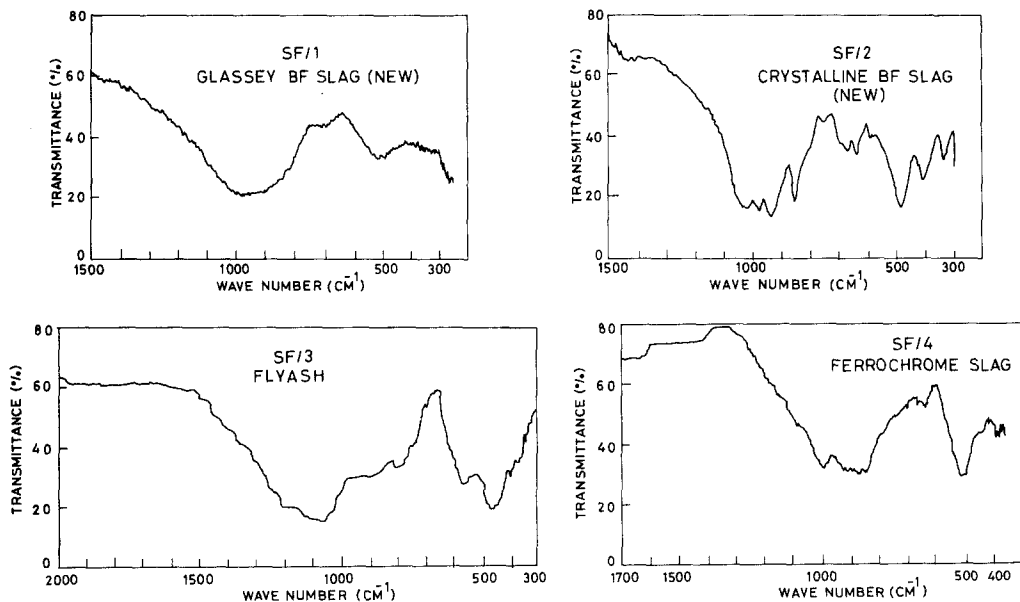


Figure 13 Spectra of slags and fly ash.

### 3.7. Spectra of biotite, muscovite, mica schist and phlogophite mica

The mica group belongs to phyllosilicate. Each  $\text{SiO}_4$  unit is linked to three other  $\text{SiO}_4$  groups similar to that in talc. The spectrum of biotite mica is characterized by strong bands at  $1018$  and  $465 \text{ cm}^{-1}$  and two weak bands at  $728$  and  $685 \text{ cm}^{-1}$ . The spectra of mica schist and muscovite are similar; the band in the  $1000 \text{ cm}^{-1}$  region is fairly broad. The spectrum of phlogophite is different to muscovite in the  $800$  to  $600 \text{ cm}^{-1}$  region. (Fig. 10)

### 3.8. Spectra of apatite, phosphatic rocks and gypsum

The spectrum of apatite rock is characterized by strong bands at  $1040$ ,  $610$  and  $570 \text{ cm}^{-1}$ , while that of gypsum is identified by the bands at  $1140$ ,  $668$  and  $602 \text{ cm}^{-1}$ . The spectrum (PS/2, Fig. 11) indicates the presence of apatite and calcite ( $1430$ ,  $876$  and  $711 \text{ cm}^{-1}$  bands) in this sample. Similarly, the spectrum (PS/3, Fig. 11) of the apatite sample, indicates the presence of quartz ( $799$ ,  $780$ ,  $690$  and  $462 \text{ cm}^{-1}$ ).

### 3.9. Spectra of red clay, high alumina, and kyanite-silimanite brick samples

The spectrum of the red clay brick sample does not indicate a large quantity of quartz. The general broad spectral feature ( $1030$  and  $460 \text{ cm}^{-1}$ ) in-

dicates the semi-crystalline nature of the material. The spectrum of the high alumina brick sample is poor. Several bands can, however, be located in its spectrum at 1150, 1100, 790, 730, 617, 555 and  $460\text{ cm}^{-1}$ . The spectrum of the kyanite-sillimanite brick sample is very complex. The bands in the  $1170\text{ cm}^{-1}$  region may be due to sillimanite [4]. Other regions of the spectrum are similar to kyanite. (Fig. 12)

### 3.9.1. Spectra of fly ash and slags

The spectrum of the fly ash sample is not well defined. The presence of bands in the region 800 to  $600\text{ cm}^{-1}$  indicates quartz and the broad band in the  $1100\text{ cm}^{-1}$  region can arise from the presence of sillimanite and mullite. The spectrum (SF/1, Fig. 13) of blast furnace slag indicates the amorphous nature of the material (broad bands) while the same slag crystallized in the laboratory shows the presence of melilite. The bands in the spectrum (SF/2, Fig. 13) closely resemble those of melilite [5]. The spectrum of the ferrochrome slag sample (SF/4, Fig. 13) shows the presence of  $\beta\text{-Ca}_2\text{SiO}_4$  [6].

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